

Low Temperature Combustion Processed Stable Al Doped ZnO Thin Film Transistor: Process Extendable up to Flexible Devices

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ABSTRACT

We report combustion synthesis of polycrystalline Aluminium doped zinc oxide (AZO) at low temperature for next generation low cost, flexible thin film transistor (TFT) application. Solution processed AZO thin film has been characterized by X ray diffraction and atomic force microscopy to confirm crystallinity. In this research work TFT with solution processed AZO as channel layer has been fabricated on both rigid and flexible substrate which exhibits excellent electrical stability and improved field effect mobility of $1.2 \text{ cm}^2\text{V}^{-1}\text{S}^{-1}$, threshold voltage of 15 V and on-off ratio of 10^6 as compared to pure ZnO based TFT. All the measurements have been carried out with varying Al concentration. Moreover, variation in defect density of AZO with Al concentration which essentially causes significant change in TFT's performance is demonstrated by chemical composition and bonding state analysis using XPS. Our results suggest that low temperature solution processed AZO TFTs have a potential for low cost, flexible and transparent electronic applications.

Keywords: Aluminium, ZnO, Combustion synthesis, Doping, Flexible Films, Thin Film Transistor (TFT), Low temperature, Spin Coating, XRD, SEM, AFM, XPS.

1. Introduction

In last few years, research on thin film transistor (TFT) is majorly focused to use transparent oxide semiconductor as channel material for future transparent display technology [1]. Zinc oxide (ZnO) is one of such semiconductor [2] and especially because of nontoxicity [3], reasonably good crystallinity [4], high electron mobility [5] and light weight [6], they are widely used for channel material in TFTs [7]. Recently several metals like In, Sn, Ga [8] have been successfully doped into pure ZnO and subsequently IZO [9] (In doped ZnO), ZTO (Sn doped ZnO) [7], IGZO (In,Ga doped ZnO) [10] have been employed as channel material in order to further improve device performance [9], [11], [12], [13]. But most of those dopants are either expensive or need process conditions such as higher temperature ($>300 \text{ }^\circ\text{C}$), high vacuum for thin

film formation which do not allow them in low cost flexible electronic applications [14]. It is because of the fact that most of the flexible substrates such as PET [15] and PEN [16] have very low glass transition temperature (T_g) ($<200 \text{ }^\circ\text{C}$). So in recent days the main challenge is to find suitable channel material and develop a low temperature device fabrication method which is appropriate for next generation low cost flexible electronics. In view of this, we choose Al doped ZnO (AZO) as channel material because Al is inexpensive and harmless to human. Recently, in few reports RF sputtered AZO [17] and PECVD deposited AZO [18] have been employed as channel layer and TFT characteristics have been subsequently investigated. But by now no work has been reported on low temperature solution process of AZO thin film.

In this article, we report combustion synthesis of Al doped ZnO, which allows the formation of

thin film of AZO on Si/SiO₂ substrate by solution spin coating and subsequent annealing at temperature down to 250 °C. Pure ZnO thin film was also prepared in similar manner for comparing performance of fabricated TFT devices. In typical experiment acetylacetone has been used here as combustion agent or fuel which provides self-generated energy to drive the reactions at low temperature [13], [11]. While fabrication we have followed staggered (top contact bottom gate) architecture and fabricated TFTs on both rigid Si and flexible PET substrate. We have got maximum field effect mobility 1.2 cm²V⁻¹S⁻¹ for 3 mol % AZO which is far better than that (0.8 cm²V⁻¹S⁻¹) of pure ZnO based TFT. Moreover, we have tried to explain the mobility variation with 3M, 5M, 7M and 9M of AZO TFT. Further we have tried to explain the mobility variation with 3M, 5M, 7M and 9M of AZO with the help of defect state analysis done by XPS measurement. Then bias stress measurement was also done to show electrical stability of the devices. So overall better performing TFTs were obtained using low temperature solution processed AZO as channel material as compared to pure ZnO.

2. Materials and Methods

All the chemicals were purchased from Sigma Aldrich and used without further purification. Firstly, zinc nitrate hexa-hydrate (Zn(HNO₃)₂·6H₂O) and acetylacetone (C₅H₈O₂) were dissolved in 5 ml of 2-methoxyethanol (C₆H₁₄O₃) to obtain 0.05 molar (M) solution. Further, Aluminium nitrate nona-hydrate (Al(NO₃)₃·9H₂O) and acetylacetone (C₅H₈O₂) were dissolved in 2-methoxyethanol to obtain 0.005M solution. Zinc nitrate hexa-hydrate and Aluminium nitrate nona-hydrate solutions were aged for 12 hr for completely dissolving metal nitrate salt in respective solvent. Combustion precursor solutions were amalgamated with variation in molar percentage (3%, 5%, 7%, and 9%) of Al (NO₃)₃·9H₂O solution to obtain AZO with variable Al mole.%. precursors with (3%, 5%, 7% and 9%). Finally, pure ZnO and different molar concentrations of AZO solutions were heated to 60 °C with magnetic stirring for 1 hr on a hot plate before film casting.

3. Device Fabrication

TFTs with top contact bottom gate structure were fabricated on both rigid (n++Si with resistivity < 0.001-0.005 ohm cm) and flexible (ITO coated PET sheet) substrate. Thermally grown SiO₂ and sputtered SiO₂ were used as dielectric on Si wafer and ITO coated PET sheet respectively. Then, on the top of this SiO₂/Si grown layer, pure ZnO and AZO combustion precursors were spin coated (Model No: SPIN NXG-P1) on SiO₂/Si substrate at 3000 rpm for 30 secs in ambience. The films were pre-heated at 150 °C for 5 mins to evaporate residual solvent and this process was repeated until the desired thickness was obtained. Finally, the samples were annealed at 250 °C for 1 hour for different concentration of the thin films. The annealing temperature is kept as 1 hour and then slowly decreased to the room temperature in ambient environment. Finally, Al source and drain contacts with channel length (L) of 100 μm and width of 600 μm were thermally evaporated using shadow mask.

4. Results and Discussion

4.1 XRD Analysis

Crystal structure was examined by X-ray diffractometry (Philips PW 3040/60) with Cu Kα radiation (λ = 1.5418 Å). We have observed the peaks corresponding to (100), (002) and (101) planes, which shows that the AZO film is polycrystalline as shown in Figure 1.

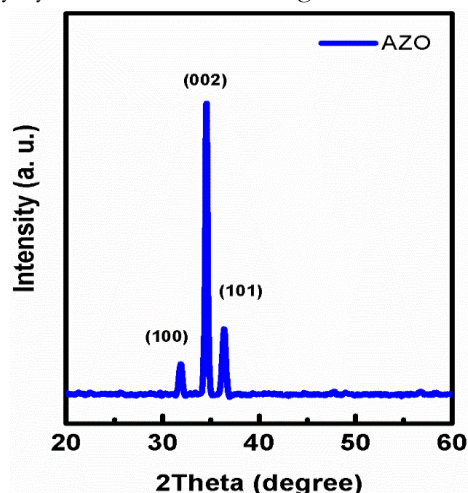


Figure 1: XRD pattern of AZO thin film deposited by spin coating technique, preheating at 150 °C and annealed at 250 °C and FEG-SEM image of AZO thin film.

The unit cell has hexagonal crystal structure with lattice parameters are $a=b=3.2498 \text{ \AA}$, $c=5.2066 \text{ \AA}$, $\alpha=\beta=90^\circ$, $\gamma=120^\circ$. Obtained peaks are compared with standard XRD pattern (JCPDS: 00-036-1451) and are normalized. It was identified that ZnO film was formed a hexagonal wurtzite structure and a preferred orientation with the c-axis perpendicular to the substrate and the particle size was calculated from the Scherrer formula as shown in equation 1.

$$D = K\lambda / (B \cos\theta) \quad (1)$$

In which the average D is the grain size and λ is the wavelength of the incident x-ray, K is a numerical constant ~ 0.96 , θ is the Bragg angle, and B is the full-width half maximum, the estimated average grain size is around $\sim 15 \text{ nm}$. XRD patterns of AZO thin films annealed at 250°C and FEG-SEM image of AZO thin film is shown in Figure 2. Thermogravimetric analysis (TGA) data is plotted for various precursors such as AZO and ZnO by using combustion method that to observed the variation of mass loss in both AZO and ZnO precursors are shown in Figure 3.

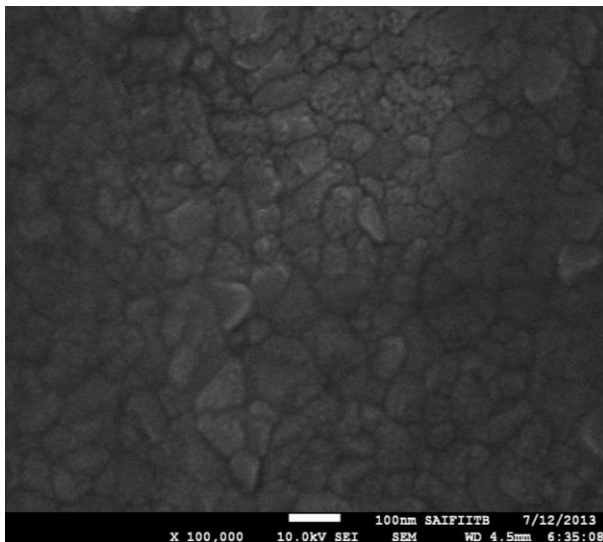


Figure 2: FEG-SEM image of AZO thin film deposited by spin coating technique, preheating at 150°C and annealed at 250°C .

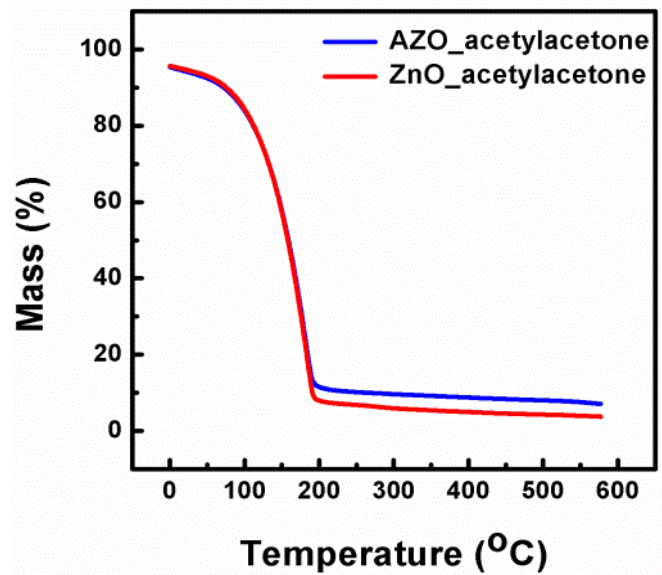


Figure 3: TGA analysis of AZO combustion and ZnO precursor solutions.

4.2 AFM Analysis

Atomic Force microscope (AFM : model MMAFMLN-AM-1897) used to measure the topography of the AZO thin films. The slightly rougher and porous surface is a common nature of solution processed AZO thin films. From AFM characterization of AZO Thin films with various molar concentration such as 3 M, 5 M, 7 M and 9 M and it is noticed that surface roughness and grain size are decreased while increased the molar concentration of AZO at annealing temperature from 250°C . It indicated that there would be more defects existing at low concentration of the AZO films with varying grain size and orientation distributions. We can see the surface roughness of the ZnO thin films in Figure 4 in 2D images.

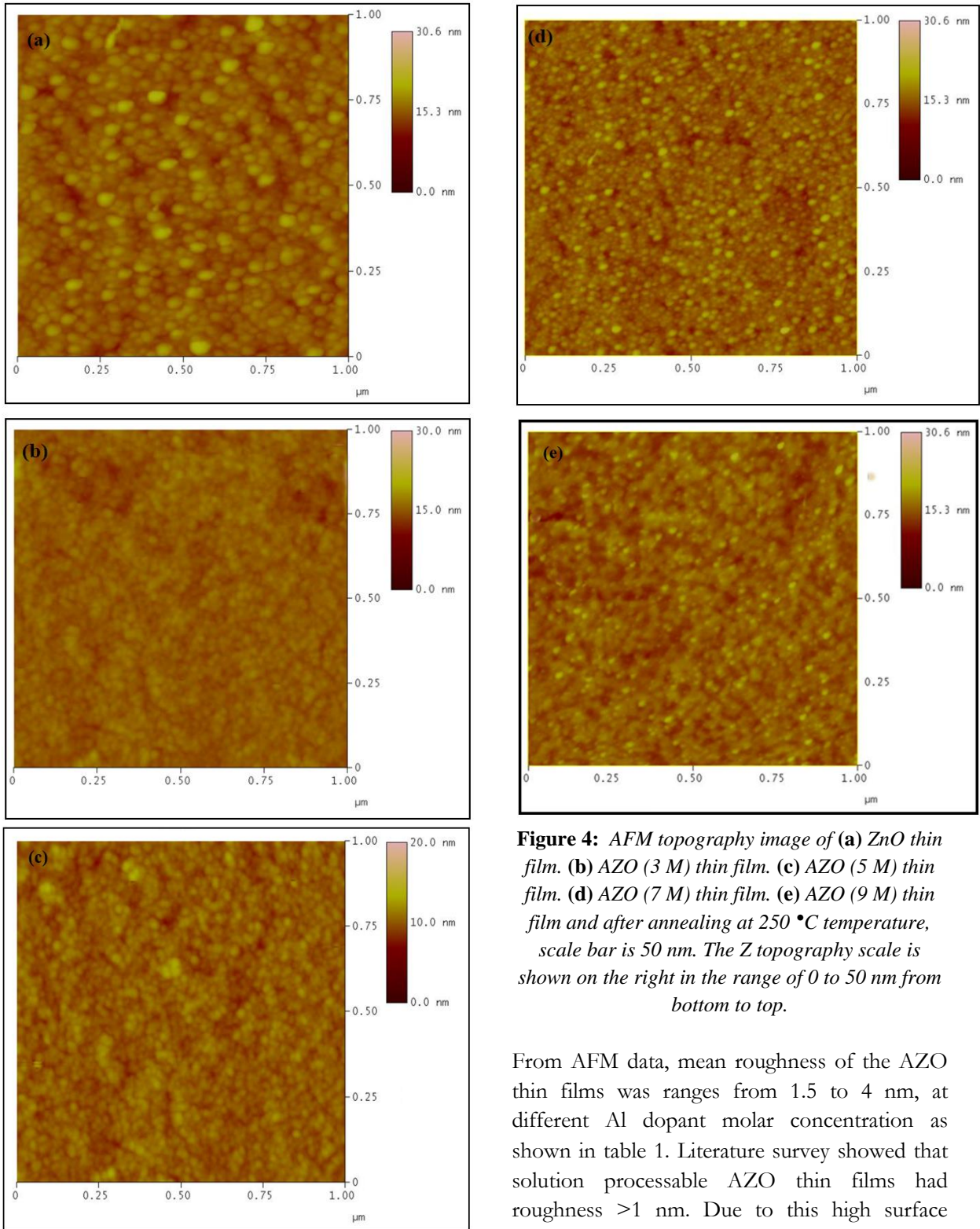


Figure 4: AFM topography image of (a) ZnO thin film. (b) AZO (3 M) thin film. (c) AZO (5 M) thin film. (d) AZO (7 M) thin film. (e) AZO (9 M) thin film and after annealing at 250 °C temperature, scale bar is 50 nm. The Z topography scale is shown on the right in the range of 0 to 50 nm from bottom to top.

From AFM data, mean roughness of the AZO thin films was ranges from 1.5 to 4 nm, at different Al dopant molar concentration as shown in table 1. Literature survey showed that solution processable AZO thin films had roughness >1 nm. Due to this high surface roughness of the AZO thin films, we obtained low mobility of AZO TFT at high Al concentration. AZO thin films surface roughness was increased with respect to decreasing of Al concentration. Thin film surface roughness can be effect on device performance and lower roughness of thin film provide the better properties of device output. The values of mean

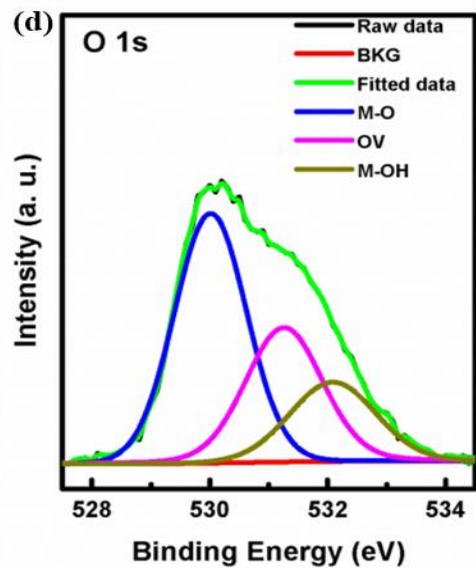
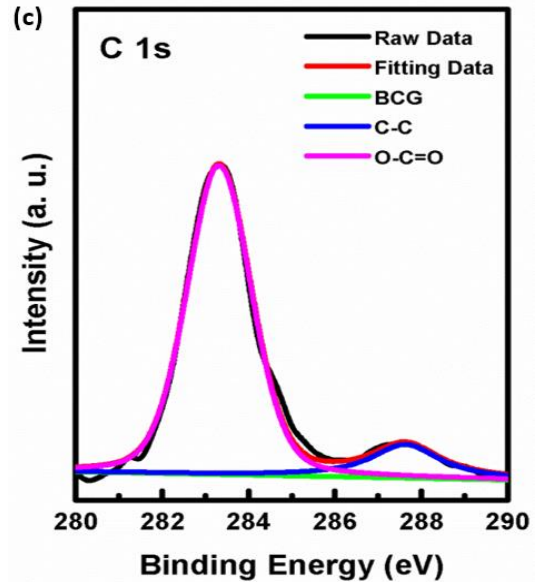
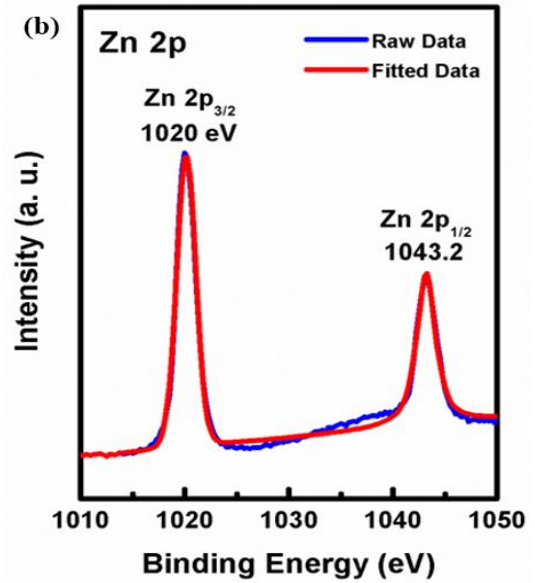
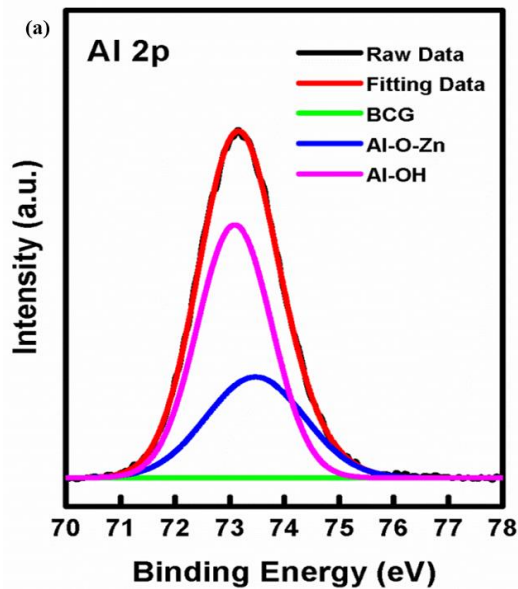
roughness at various concentration Al of the AZO thin films were shown in Table 1.

Table 1: Surface roughness for AZO thin films of different Al dopant molar concentrations

Material	Molar concentration	Mean roughness (nm)
ZnO thin film	0	4
AZO thin film	3	3.7
AZO thin film	5	1.8
AZO thin film	7	1.5
AZO thin film	9	1

4.3 XPS Analysis

X-ray photoelectron spectroscopy (XPS) was used to analyse AZO surface impurities, carbon contamination, and oxygen binding as shown in Figures 5 below.



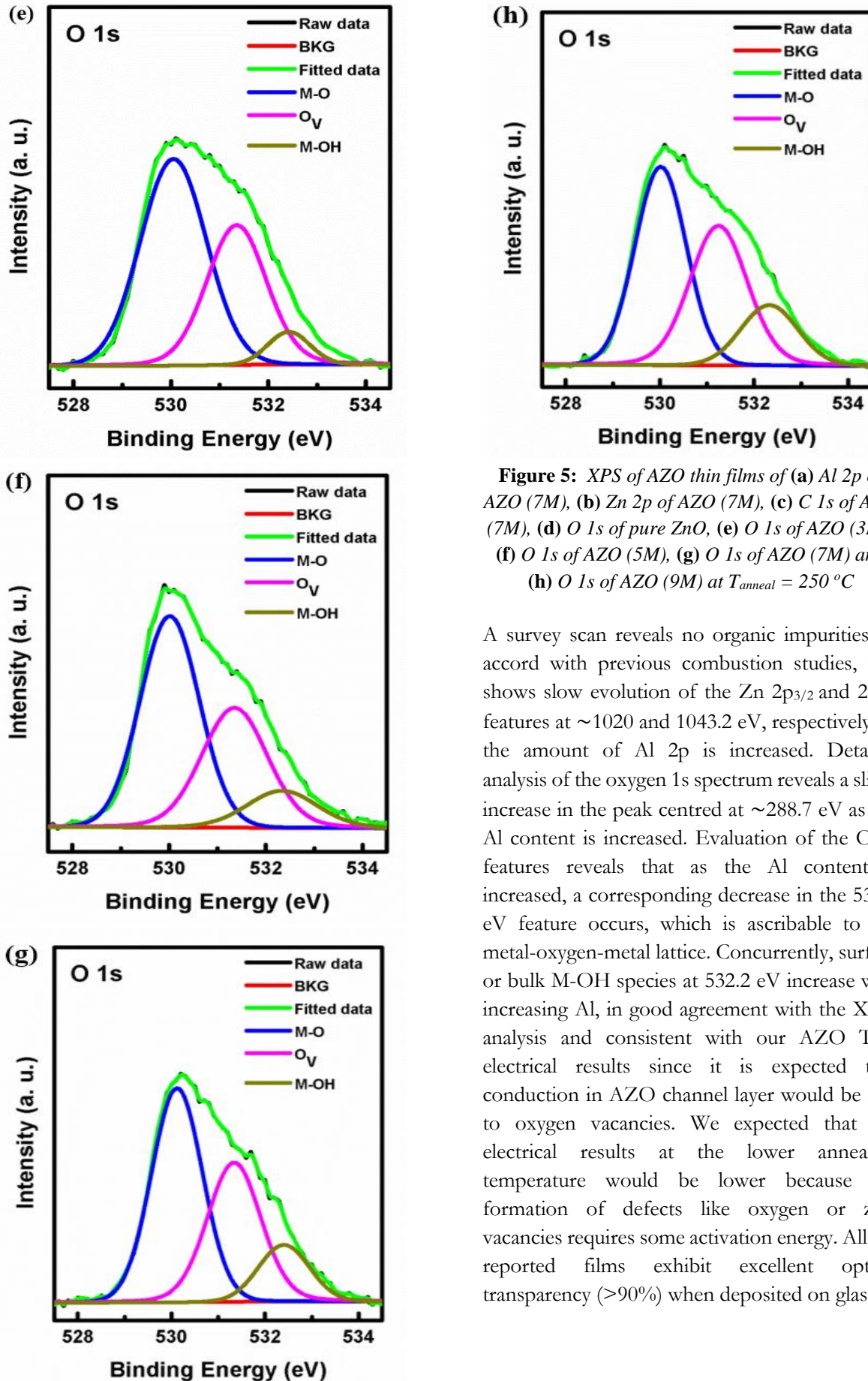


Figure 5: XPS of AZO thin films of (a) Al 2p of AZO (7M), (b) Zn 2p of AZO (7M), (c) C 1s of AZO (7M), (d) O 1s of pure ZnO, (e) O 1s of AZO (3M), (f) O 1s of AZO (5M), (g) O 1s of AZO (7M) and (h) O 1s of AZO (9M) at $T_{anneal} = 250\text{ }^{\circ}\text{C}$

A survey scan reveals no organic impurities, in accord with previous combustion studies, and shows slow evolution of the Zn 2p_{3/2} and 2p_{1/2} features at ~1020 and 1043.2 eV, respectively, as the amount of Al 2p is increased. Detailed analysis of the oxygen 1s spectrum reveals a slight increase in the peak centred at ~288.7 eV as the Al content is increased. Evaluation of the O 1s features reveals that as the Al content is increased, a corresponding decrease in the 530.7 eV feature occurs, which is ascribable to the metal-oxygen-metal lattice. Concurrently, surface or bulk M-OH species at 532.2 eV increase with increasing Al, in good agreement with the XRD analysis and consistent with our AZO TFT electrical results since it is expected that conduction in AZO channel layer would be due to oxygen vacancies. We expected that the electrical results at the lower annealing temperature would be lower because the formation of defects like oxygen or zinc vacancies requires some activation energy. All the reported films exhibit excellent optical transparency (>90%) when deposited on glass.

4.4 Thin Film Transistors Electrical Characterization

To investigate electrical performances of as prepared AZO thin film, a TFT was fabricated following bottom gate top contact architecture. Here SiO₂ and thermally evaporated Al were used for gate dielectric and source/drain contacts respectively. A schematic of the device structure of flexible TFTs has been shown in Figure 6.

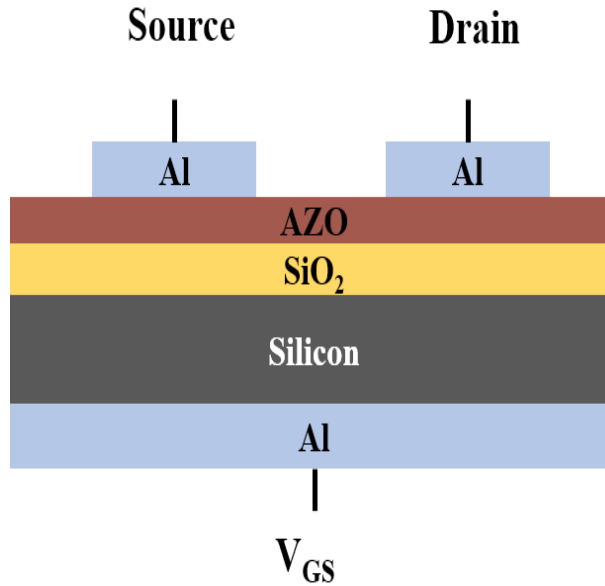


Figure 6: Schematic of the bottom gate AZO Thin film transistor.

4.4.1 Field Effect Mobility Extraction

In prevalent method, for crystalline MOSFETs with a grounded source, field-effect mobility is extracted by using the following expression that describes the drain current (I_D) is shown in equation 2. [15].

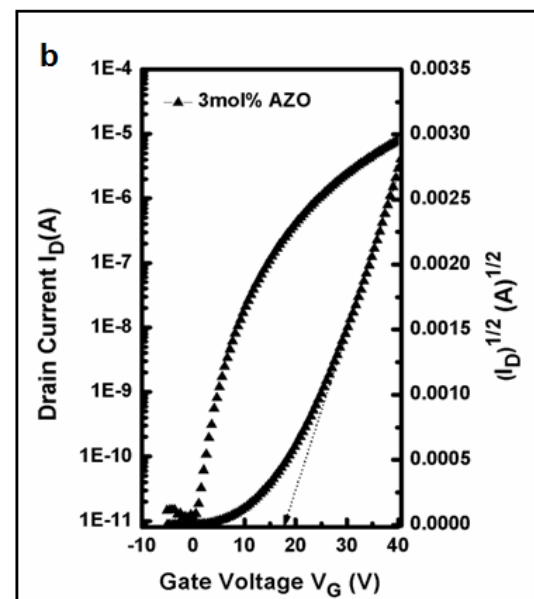
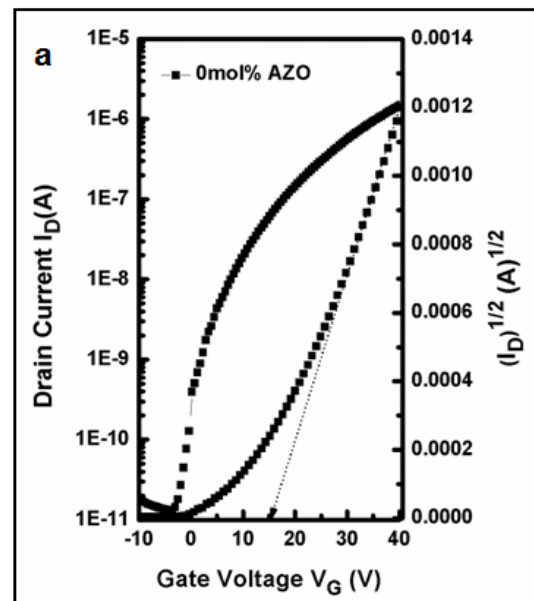
$$I_D = \frac{W}{2L} \mu_{sat} C_{ox} (V_G - V_T)V_D - \frac{V_D^2}{2} \quad (2)$$

Where V_D is drain-source voltage, V_G is (gate-source voltage, μ is the field-effect mobility, V_T is the threshold voltage, and C_{ox} is the capacitance per unit area of the gate insulator. In saturation region is $V_D \geq V_G - V_T$. So eqn. 1 can be simplified equation that is shown in equation 3.

$$I_{D,sat} = \frac{W}{2L} \mu_{sat} C_{ox} (V_G - V_T)^2 \quad (3)$$

Field experimental details and I_D - V_G plots are shows the typical transfer characteristics of AZO TFTs performance metrics that respectively for

varied Al doping concentrations for $T_{anneal} = 250^\circ\text{C}$ are shown in Figure 7. The device performed excellent properties, such as a high ON/OFF ratio of 10^9 at threshold of 15 V, low mobility μ_s of $0.08 \text{ cm}^2/\text{V}\cdot\text{s}$ for pure ZnO TFT. At lower growth temperatures, a maximum mobility of $1.2 \text{ cm}^2 \text{ V}^{-1}\text{s}^{-1}$ is achieved with a large current on/off ratio of 10^6 and a threshold voltage (V_T) of 17 V. Notably, as the Al doping is increased, the mobility decreases significantly, and the devices become inactive at $\geq 7\text{mol} \%$ Al. All the calculated AZO TFT mobilities are shown in table 2.



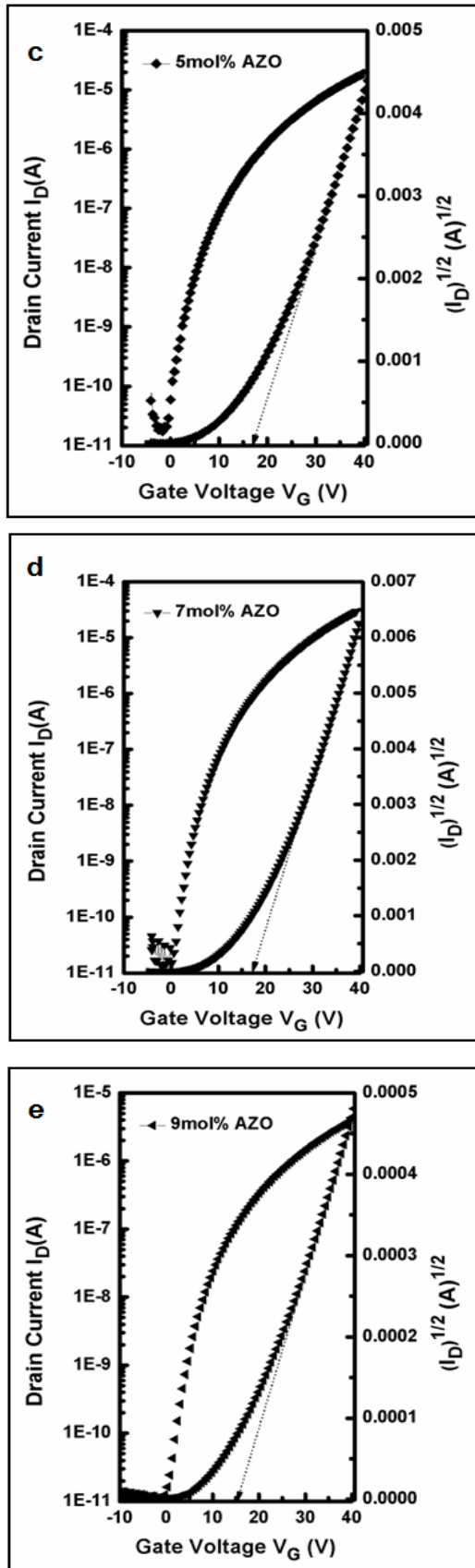


Figure 7: *I-V* measurement for TFT of different Al dopant molar concentrations (a) ZnO, (b) AZO with 3M, (c) AZO with 5M, (d) AZO with 7M and (e) AZO with 9M.

Moreover, as the Al content is increased from 3M to 9M, the oxygen vacancy and free carrier concentrations decrease. As a result, I_{on}/I_{off} initially increases because of the reduced I_{off} , while a similar I_{on} is maintained. Further increases in Al content depress I_{on}/I_{off} as a result of the significant decrease in I_{on} from the mobility degradation. On the other hand, V_T is related to the free carrier concentration and trap density. At low Al content, V_T follows the trend expected from the decrease in carrier concentration, further confirming the role of oxygen vacancy suppression.

Table 2: *IV* measurement for AZO TFT of different Al dopant molar concentrations.

ZnO TFT and doped Al	Mobility $\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$	Threshold voltage (V_T)	Sub threshold	I_{on}/I_{off}
ZnO, 0M	0.08	15	10.15	$\sim 10^6$
AZO, 3M	0.5	17	7.7	$\sim 10^6$
AZO, 5M	0.9	16	5.2	$\sim 10^6$
AZO, 7M	1.2	17	4.5	$\sim 10^6$
AZO, 9M	0.85	15	6.1	$\sim 10^6$

However, as the Al content is increased, we observe an unstable trend in V_T . Moreover, when increasing the Al doping concentration in ZnO that would be effect on electrons density per unit volume of AZO to be increased. AZO (7M) is optimum one because of this molar concentration provides the moderate electrons to conduct in the channel layers. Where, AZO (9M) or higher concentration is providing the over electrons to conduct in the channel that causes the scattering effect in the electrons cloud and impurities incorporation in substitutional sites would be increased consequently device performance will be poor at higher molar concentration.

4.4.2 Thin-Film Transistor Bias Stress Properties

To be useful in TFTs, an effective semiconducting channel material must be stable under constant bias stress [19]. Previous reports have shown that metal oxide semiconductor performance under bias stress is related to a variety of factors, including H₂O and O₂ adsorption on the back channel and electron trapping at the semiconductor/dielectric interface. Thin semiconductor films (as in the present case), H₂O adsorption has been shown to create an accumulation layer due to electron donation, resulting in a negative V_T shift. On the other hand, O₂ adsorption is known to form a depletion layer below the active surface, leading to a positive V_T shift. Unlike trapping at the semiconductor/dielectric interface, these two factors can be mitigated by appropriate passivation of the channel. Measured the properties of the AZO channel materials, TFTs were fabricated as described previously (here channel L = 200, W = 300 μm) with 7M concentration of Al was the better performance of the device and T_{anneal} = 250 °C. These devices were then subjected to a V_G-V_D constant bias of +10 V for 100 s intervals for the duration of 1000 s in ambient, with intentional light blocking. The resulting transfer plots are shown in Figure 8.

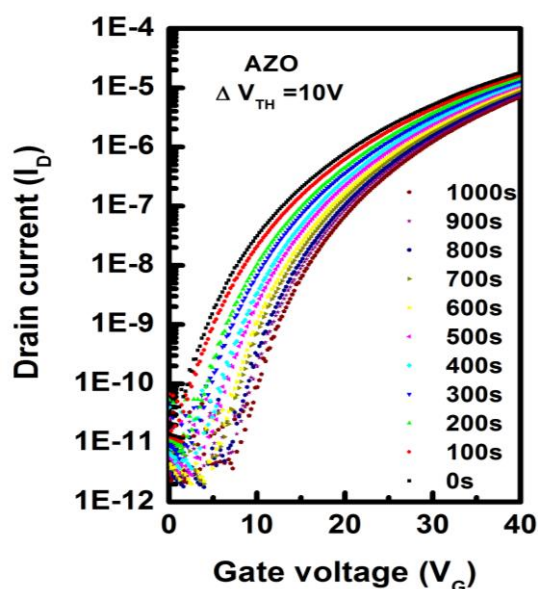


Figure 8: Effect of the gate bias stress on the transfer characteristics for AZO TFTs with Al dopant of 7M and T_{anneal} = 250 °C. Gate-to-drain bias +10 V was maintained for 100 s between transfer measurements for 1000 s.

As expected, the direction of bias related shifts are toward positive voltages as indicated by the black arrow in the top left pane. This positive shift is expected from previous studies which argued that under positive bias stress a AZO forms oxygen-related electron-trapping (acceptor-like) states. All four AZO TFT classes exhibit a fall in I_{off} over the bias duration, although the fall is significantly more limited in the case of AZO devices. The threshold voltage for each system lies within the typical range observed for AZO TFTs investigated in this study. In general, the best-performing TFTs having films prepared with each oxygen getter at both film processing temperatures exhibit V_T's in the range of 0 to 10V, with a positive shift observed with increasing Al concentration up to 7M.

5. Conclusions

We demonstrated, combustion synthesis of polycrystalline Aluminium doped zinc oxide (AZO) thin film transistors at low annealing temperatures of 250 °C with a maximum field effect mobility value of 1.2 cm² V⁻¹ s⁻¹ and threshold voltage of 17 V at 7 mol. % of Al doping. Devices without Al doping demonstrated field effect mobility of 0.08 cm² V⁻¹ s⁻¹ and 15 V, respectively. Based on the obtained results, it is confirmed that Al act as a very effective dopant for enhancing the electrical performance at 250 °C and device stability of ZnO based TFTs by increasing charge carrier concentration and reducing the number of scattering centres. Thus, combustion synthesis of polycrystalline Aluminium doped zinc oxide (AZO) thin film transistors may enable several low cost flexible electronic applications.

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